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## 4-Phenyl-6,7,8,9-tetrahydro-[1]benzothieno-[3,2-e][1,2,4]triazolo[4,3-a]pyrimidin-5(4H)one

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#### Abstract

The cyclohexane ring of the title compound, $\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{~N}_{4}-$ OS, shows disorder with an $85 \%$ major conformer and a $15 \%$ minor conformer. Both adopt half-chair conformations. The phenyl ring is tilted by $80.79(4)^{\circ}$ to the plane formed by the rest of the atoms of the mol-

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ecule. Molecules of the same orientation are linked by $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds and short $\mathrm{S} \cdots \mathrm{O}$ contacts, forming chains along the $b$ axis.

## Comment

Condensed [1,2,4]triazoles are biologically important compounds (Kottke, Kuesmstedt, Hagen, Renner \& Schnitzler, 1983; Francis \& Gelotte, 1988; Francis et al., 1988). Since our interest is in the synthesis and characterization of biologically active heterocyclic compounds, especially thienopyrimidines, we have prepared a series of [4,3-a]thieno[1,2,4]triazolo[3,2-e]pyrimidin$5(4 H)$-one derivatives and studied both their depressant effect on the central nervous system and their analgesic activities (Pathak, Gandhi, Singh, Warde \& Jain, 1992). The structure of the title compound, (I), one of these derivatives, has been determined in order to elucidate the conformational features of the molecule. To our knowledge, this is the first crystal structure report of a [ $1,2,4$ ]triazolo- or thienopyrimidine derivative.

(I)

The bond lengths and angles in the title compound have normal values. The part of the molecule comprising the thiene, pyrimidine and triazo rings is planar, with a maximum deviation of 0.050 (2) $\AA$ for atom C 15 . Atoms C2 and C5 also lie close to this plane and the phenyl ring is almost perpendicular to it [dihedral angle


Fig. 1. The structure of the title compound showing $50 \%$ probability displacement ellipsoids and the atom-numbering scheme. The minor conformer positions are linked by open bonds.
$\left.80.79(4)^{\circ}\right]$. Atoms C3 and C4 of the cyclohexane ring (C1-C6) are each disordered over two positions, with 85 and $15 \%$ occupancies. The two alternative conformations of this ring are both half chair and differ in the direction of the displacement of atoms C 3 and C 4 from the plane defined by $\mathrm{C} 1, \mathrm{C} 2, \mathrm{C} 5$ and C 6 (relevant torsion angles are given in Table 2). The carbonyl O22 atom is involved in a $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bond and a short $\mathrm{S} \cdots \mathrm{O}$ contact with a neighbouring molecule [ $\mathrm{C} 2-\mathrm{H} 2 \mathrm{~A}$ 0.98 (3), $\mathrm{H} 2 \mathrm{~A} \cdots \mathrm{O} 22^{\mathrm{i}} 2.52(2), \mathrm{C} 2 \cdots \mathrm{O} 22^{\mathrm{i}} 3.374$ (3) $\AA$ and C2-H2A‥O22 $146(2)^{\circ} ; \mathrm{S} \cdots 2^{\mathrm{i}} 3.189$ (2) $\AA$; symmetry code: (i) $\left.\frac{1}{2}-x, y-\frac{1}{2}, z\right]$. These two contacts link molecules of the same orientation into chains along the $b$ axis, as shown in Fig. 2.


Fig. 2. Packing of the molecules viewed down the $a$ axis, with the C $\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds and short $\mathrm{S} \cdots \mathrm{O}$ contacts shown by dashed lines.

## Experimental

Details of the synthesis of the title compound are published elsewhere (Pathak, Gandhi, Singh, Warde \& Jain, 1992). Single crystals were obtained by recrystallization from an ethanol-chloroform mixture.

## Crystal data

$\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{OS}$
$M_{r}=322.38$
Orthorhombic
Pbca
$a=10.668$ (1) $\AA$
$b=14.687(2) \AA$
$c=18.820(2) \AA$
$V=2948.7(6) \AA^{3}$
$Z=8$
$D_{x}=1.452 \mathrm{Mg} \mathrm{m}^{-3}$

Mo $K \alpha$ radiation
$\lambda=0.71073 \AA$
Cell parameters from 25 reflections
$\theta=8-25^{\circ}$
$\mu=0.230 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Rectangular slab
$0.66 \times 0.52 \times 0.26 \mathrm{~mm}$ Transparent

## Data collection

Siemens $P 4$ diffractometer
$\theta / 2 \theta$ scans
Absorption correction: none
3270 measured reflections
2560 independent reflections
2034 observed reflections
$[I>2 \sigma(I)]$
$R_{\text {int }}=0.0193$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.0341$
$w R\left(F^{2}\right)=0.0996$
$S=1.037$
2560 reflections
271 parameters
All H -atom parameters
refined

$$
\begin{aligned}
& w= 1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0582 P)^{2}\right. \\
&+0.1938 P] \\
& \text { where } P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3
\end{aligned}
$$

$\theta_{\text {max }}=24.99^{\circ}$
$h=-1 \rightarrow 12$
$k=-1 \rightarrow 17$
$l=-1 \rightarrow 22$
3 standard reflections monitored every 100 reflections
intensity decay: <4\%
$(\Delta / \sigma)_{\text {max }}<0.001$
$\Delta \rho_{\text {max }}=0.22 \mathrm{e} \AA^{-3}$
$\Delta \rho_{\text {min }}=-0.22 \mathrm{e}^{-3}$
Atomic scattering factors from International Tables for Crystallography (1992, Vol. C, Tables 4.2.6.8 and 6.1.1.4)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters $\left(\AA^{2}\right)$

|  | $x$ | $y$ | $z$ | $U_{\text {eq }}$ |
| :---: | :---: | :---: | :---: | :---: |
| S | 0.09195 (4) | 0.84896 (3) | 0.01955 (2) | 0.0403 (2) |
| C1 | 0.2124 (2) | 0.86993 (12) | -0.04086 (9) | 0.0384 (4) |
| C2 | 0.2512 (2) | 0.79996 (14) | -0.09487 (12) | 0.0495 (5) |
| C3 $\dagger$ | 0.3777 (3) | 0.8243 (2) | -0.1258 (2) | 0.0529 (8) |
| C3' $\dagger$ | 0.3400 (19) | 0.8467 (13) | -0.1498 (10) | 0.0529 (8) |
| C4 $\dagger$ | 0.3862 (3) | 0.9243 (2) | -0.1444 (2) | 0.0508 (8) |
| C4' $\dagger$ | 0.4378 (17) | 0.9049 (12) | -0.1142 (9) | 0.0508 (8) |
| C5 | 0.3697 (2) | 0.98631 (15) | -0.07948 (11) | 0.0454 (5) |
| C6 | 0.2643 (2) | 0.95364 (12) | -0.03269 (9) | 0.0359 (4) |
| C7 | 0.2056 (2) | 1.00408 (12) | 0.02394 (8) | 0.0333 (4) |
| C8 | 0.2411 (2) | 1.09378 (12) | 0.05058 (9) | 0.0366 (4) |
| N9 | 0.17003 (14) | 1.12454 (10) | 0.10892 (7) | 0.0360 (3) |
| C10 | 0.0741 (2) | 1.07456 (12) | 0.13839 (9) | 0.0345 (4) |
| N11 | 0.00326 (14) | 1.09584 (11) | 0.19227 (8) | 0.0425 (4) |
| N12 | -0.07673 (15) | 1.02130 (11) | 0.20233 (8) | 0.0466 (4) |
| C13 | -0.0502 (2) | 0.96061 (13) | 0.15462 (10) | 0.0411 (4) |
| N14 | 0.04470 (13) | 0.99058 (9) | 0.11158 (7) | 0.0339 (3) |
| C15 | 0.1117 (2) | 0.95512 (11) | 0.05486 (9) | 0.0333 (4) |
| C16 | 0.2016 (2) | 1.21037 (12) | 0.14291 (9) | 0.0358 (4) |
| C17 | 0.2991 (2) | 1.21260 (14) | 0.19048 (10) | 0.0442 (5) |
| C18 | 0.3274 (2) | 1.2942 (2) | 0.22400 (11) | 0.0553 (6) |
| C19 | 0.2571 (2) | 1.3707 (2) | 0.21073 (12) | 0.0580 (6) |
| C20 | 0.1597 (2) | 1.36710 (14) | 0.15364 (14) | 0.0600 (6) |
| C21 | 0.1308 (2) | 1.28684 (14) | 0.12834 (12) | 0.0490 (5) |
| 022 | 0.32536 (14) | 1.14048 (9) | 0.02664 (7) | 0.0527 (4) |

## $\dagger$ Partially occupied (see below).

Table 2. Selected geometric parameters ( $\AA,{ }^{\circ}$ )

| $\mathrm{S}-\mathrm{C} 15$ | $1.708(2)$ | $\mathrm{C} 8-\mathrm{N} 9$ | $1.409(2)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{S}-\mathrm{C} 1$ | $1.743(2)$ | $\mathrm{N} 9-\mathrm{C} 10$ | $1.376(2)$ |
| $\mathrm{C} 1-\mathrm{C} 6$ | $1.357(3)$ | $\mathrm{N} 9-\mathrm{C} 16$ | $1.453(2)$ |
| $\mathrm{C} 1-\mathrm{C} 2$ | $1.504(3)$ | $\mathrm{C} 10-\mathrm{N} 11$ | $1.302(2)$ |
| $\mathrm{C} 2-\mathrm{C} 3$ | $1.513(4)$ | $\mathrm{C} 10-\mathrm{N} 14$ | $1.369(2)$ |
| $\mathrm{C} 2-\mathrm{C} 3^{\prime}$ | $1.56(2)$ | $\mathrm{N} 11-\mathrm{N} 12$ | $1.401(2)$ |
| $\mathrm{C} 3-\mathrm{C} 4$ | $1.512(5)$ | $\mathrm{N} 12-\mathrm{C} 13$ | $1.296(2)$ |
| $\mathrm{C} 3^{\prime}-\mathrm{C} 4^{\prime}$ | $1.51(3)$ | $\mathrm{C} 13-\mathrm{N} 14$ | $1.370(2)$ |
| $\mathrm{C} 4-\mathrm{C} 5$ | $1.534(3)$ | $\mathrm{N} 14-\mathrm{C} 15$ | $1.386(2)$ |
| $\mathrm{C} 4^{\prime}-\mathrm{C} 5$ | $1.54(2)$ | $\mathrm{C} 16-\mathrm{C} 17$ | $1.373(3)$ |
| $\mathrm{C} 5-\mathrm{C} 6$ | $1.506(3)$ | $\mathrm{C} 16-\mathrm{C} 21$ | $1.381(3)$ |

$$
\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{OS}
$$

| C6-C7 | 1.441 (2) | C17-C18 | 1.388 (3) |
| :---: | :---: | :---: | :---: |
| C7-C15 | 1.364 (2) | C18-C19 | 1.373 (4) |
| C7-C8 | 1.460 (2) | C19-C20 | 1.367 (4) |
| C8-022 | 1.217 (2) | C20-C21 | 1.388 (3) |
| C15-S-Cl | 90.10 (8) | C10-N9-C16 | 117.23 (14) |
| $\mathrm{C} 6-\mathrm{Cl}-\mathrm{C} 2$ | 125.7 (2) | C8-N9-C16 | 119.77 (14) |
| C6-C1-S | 112.76 (13) | N11-C10-N14 | 111.7 (2) |
| $\mathrm{C} 2-\mathrm{Cl}-\mathrm{S}$ | 121.54 (15) | N11-C10-N9 | 128.1 (2) |
| $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ | 110.1 (2) | N14-C10-N9 | 120.14 (15) |
| $\mathrm{Cl}-\mathrm{C} 2-\mathrm{C}^{\prime}$ | 108.3 (7) | C10-N11-N12 | 105.71 (14) |
| C4-C3-C2 | 111.8 (3) | C13-N12-N11 | 108.12 (15) |
| $\mathrm{C} 4^{\prime}-\mathrm{C} 3^{\prime}-\mathrm{C} 2$ | 112.1 (14) | N12-C13-N14 | 110.5 (2) |
| C3-C4-C5 | 112.7 (2) | C10-N14-C13 | 103.93 (14) |
| $\mathrm{C} 3^{\prime}-\mathrm{C} 4^{\prime}-\mathrm{C} 5$ | 107.6 (14) | C10-N14-C15 | 120.30 (14) |
| C6-C5-C4 | 111.2 (2) | C13-N14-C15 | 135.7 (2) |
| C6-C5-C4 ${ }^{\prime}$ | 110.6 (6) | C7-C15-N14 | 120.6 (2) |
| $\mathrm{Cl}-\mathrm{C} 6-\mathrm{C} 7$ | 111.9 (2) | C7-C15-S | 113.94 (13) |
| C1-C6-C5 | 121.8 (2) | N14-C15-S | 125.40 (13) |
| C7-C6-C5 | 126.4 (2) | C17-C16-C21 | 121.6 (2) |
| C15-C7-C6 | 111.3 (2) | C17-C16-N9 | 118.9 (2) |
| C15-C7-C8 | 121.3 (2) | C21-C16-N9 | 119.4 (2) |
| C6-C7-C8 | 127.3 (2) | C16-C17-C18 | 118.8 (2) |
| O22-C8-N9 | 120.4 (2) | C19-C18-C17 | 120.3 (2) |
| O22-C8-C7 | 125.0 (2) | C20-C19-C18 | 120.1 (2) |
| N9-C8-C7 | 114.67 (15) | C19-C20-C21 | 120.8 (2) |
| C10-N9-C8 | 122.92 (15) | C16-C21-C20 | 118.3 (2) |
| $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4$ | -44.7 (3) | $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3^{\prime}-\mathrm{C}^{\prime}$ | 45.6 (17) |
| C2-C3-C4-C5 | 61.2 (4) | $\mathrm{C} 2-\mathrm{C}^{\prime}-\mathrm{C} 4^{\prime}-\mathrm{C} 5$ | -67.5 (19) |
| C3-C4-C5-C6 | -43.0 (4) | $\mathrm{C} 3^{\prime}-\mathrm{C} 4^{\prime}-\mathrm{C} 5-\mathrm{C} 6$ | 52.5 (16) |
| C4-C5-C6--Cl | 13.0 (3) | C4'- 5 - 6 C $6-\mathrm{Cl}$ | -21.0 (8) |
| C5- $66-\mathrm{Cl}-\mathrm{C} 2$ | 0.6 (3) | C6- $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C}^{\prime}$ | -12.0 (9) |
| C6-C1-C2-C3 | 15.3 (3) |  |  |

Disorder of atoms C3 and C4 was observed after anisotropic refinement of the non- H atoms. C 3 and C 4 were assigned occupancies of $p$ and $\mathrm{C} 3^{\prime}$ and $\mathrm{C} 4^{\prime}$ were assigned occupancies of $1-p$, with $p=0.850(6)$ at convergence. All the H atoms belonging to the major conformer were located from difference Fourier maps and refined isotropically.

Data collection: XSCANS (Siemens, 1994). Cell refinement: XSCANS. Data reduction: XSCANS. Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990a). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: SHELXTL/PC (Sheldrick, 1990b). Software used to prepare material for publication: SHELXL93. Geometric calculations: PARST (Nardelli, 1983).

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Lists of structure factors, anisotropic displacement parameters, Hatom coordinates and complete geometry have been deposited with the IUCr (Reference: MU1178). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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## Abstract

Crystals of the title compound, $\mathrm{C}_{15} \mathrm{H}_{10} \mathrm{O}_{2}$, are monoclinic, space group $P 2_{1} / c$. The molecules are disordered around a centre of symmetry. Packing considerations and the calculation of the intermolecular interaction energies show that the disordered structure has occurred by random packing of structures with space groups $P 2_{1}$ and $P c$, the maximal subgroups of $P 2_{1} / c$.

## Comment

A certain kind of structural disorder has been observed in homomolecular crystals like those of azulene (Robertson, Shearer, Sim \& Watson, 1962) and 1-bromo-4chlorobenzene (Klug, 1947). In these crystals, each molecule is apparently disordered in two orientations related by a $\overline{1}$ symmetry operation. Here, we comment on such a structural disorder as found in 2-methylanthraquinone, (I).


The molecular arrangement is characterized by the formation of a molecular sheet in (10 $\overline{2}$ ) and molecular stacking along the $a$ axis with an interplanar spacing

