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ecule. Molecules of the same orientation are linked by C—H···O hydrogen bonds and short S···O contacts, forming chains along the b axis.

## Comment

Condensed [1,2,4]triazoles are biologically important compounds (Kottke, Kuesmstedt, Hagen, Renner & Schnitzler, 1983; Francis & Gelotte, 1988; Francis *et al.*, 1988). Since our interest is in the synthesis and characterization of biologically active heterocyclic compounds, especially thienopyrimidines, we have prepared a series of [4,3-*a*]thieno[1,2,4]triazolo[3,2-*e*]pyrimidin-5(4H)-one derivatives and studied both their depressant effect on the central nervous system and their analgesic activities (Pathak, Gandhi, Singh, Warde & Jain, 1992). The structure of the title compound, (I), one of these derivatives, has been determined in order to elucidate the conformational features of the molecule. To our knowledge, this is the first crystal structure report of a [1,2,4]triazolo- or thienopyrimidine derivative.



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# 4-Phenyl-6,7,8,9-tetrahydro-[1]benzothieno-[3,2-*e*][1,2,4]triazolo[4,3-*a*]pyrimidin-5(4*H*)one

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# Abstract

The cyclohexane ring of the title compound,  $C_{17}H_{14}N_4$ -OS, shows disorder with an 85% major conformer and a 15% minor conformer. Both adopt half-chair conformations. The phenyl ring is tilted by 80.79 (4)° to the plane formed by the rest of the atoms of the molThe bond lengths and angles in the title compound have normal values. The part of the molecule comprising the thiene, pyrimidine and triazo rings is planar, with a maximum deviation of 0.050 (2) Å for atom C15. Atoms C2 and C5 also lie close to this plane and the phenyl ring is almost perpendicular to it [dihedral angle



Fig. 1. The structure of the title compound showing 50% probability displacement ellipsoids and the atom-numbering scheme. The minor conformer positions are linked by open bonds.

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80.79 (4)°]. Atoms C3 and C4 of the cyclohexane ring (C1–C6) are each disordered over two positions, with 85 and 15% occupancies. The two alternative conformations of this ring are both half chair and differ in the direction of the displacement of atoms C3 and C4 from the plane defined by C1, C2, C5 and C6 (relevant torsion angles are given in Table 2). The carbonyl O22 atom is involved in a C—H···O hydrogen bond and a short S···O contact with a neighbouring molecule [C2—H2A 0.98 (3), H2A···O22<sup>i</sup> 2.52 (2), C2···O22<sup>i</sup> 3.374 (3) Å and C2—H2A···O22<sup>i</sup> 146 (2)°; S···O22<sup>i</sup> 3.189 (2) Å; symmetry code: (i)  $\frac{1}{2} - x, y - \frac{1}{2}, z$ ]. These two contacts link molecules of the same orientation into chains along the *b* axis, as shown in Fig. 2.



Fig. 2. Packing of the molecules viewed down the *a* axis, with the C— $H \cdots O$  hydrogen bonds and short  $S \cdots O$  contacts shown by dashed lines.

#### Experimental

Details of the synthesis of the title compound are published elsewhere (Pathak, Gandhi, Singh, Warde & Jain, 1992). Single crystals were obtained by recrystallization from an ethanol-chloroform mixture.

#### Crystal data

$C_{17}H_{14}N_4OS$	Mo $K\alpha$ radiation
$M_r = 322.38$	$\lambda = 0.71073 \text{ Å}$
Orthorhombic	Cell parameters from 25
Pbca	reflections
a = 10.668 (1)  Å	$\theta = 8-25^{\circ}$
b = 14.687 (2) Å	$\mu = 0.230 \text{ mm}^{-1}$
c = 18.820(2) Å	T = 293 (2)  K
$V = 2948.7 (6) Å^3$	Rectangular slab
Z = 8	$0.66 \times 0.52 \times 0.26$ mm
$D_x = 1.452 \text{ Mg m}^{-3}$	Transparent
_	

# Data collection

Siemens P4 diffractometer	$\theta_{\rm max} = 24.99^{\circ}$
$\theta/2\theta$ scans	$h = -1 \rightarrow 12$
Absorption correction:	$k = -1 \rightarrow 17$
none	$l = -1 \rightarrow 22$
3270 measured reflections	3 standard reflections
2560 independent reflections	monitored every 100
2034 observed reflections	reflections
$[I > 2\sigma(I)]$	intensity decay: <4%
$R_{\rm int} = 0.0193$	

#### Refinement

N12

C13

N14 C15 C16 C17

C18 C19

C20

C21

022

Refinement on $F^2$
$R[F^2 > 2\sigma(F^2)] = 0.0341$
$wR(F^2) = 0.0996$
S = 1.037
2560 reflections
271 parameters
All H-atom parameters
refined
$w = 1/[\sigma^2(F_o^2) + (0.0582P)^2]$
+ 0.1938 <i>P</i> ]
where $P = (F_0^2 + 2F_c^2)/3$

#### $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.22 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{min} = -0.22 \text{ e} \text{ Å}^{-3}$ Atomic scattering factors from International Tables for Crystallography (1992, Vol. C, Tables 4.2.6.8 and 6.1.1.4)

# Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å<sup>2</sup>)

$U_{eq} = (1)$	$(3)\Sigma_i\Sigma_jU$	$_{ij}a_i^*a_i^*\mathbf{a}_i.\mathbf{a}_j.$
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		•	
x	у	Z	$U_{eq}$
0.09195 (4)	0.84896 (3)	0.01955 (2)	0.0403 (2)
0.2124 (2)	0.86993 (12)	-0.04086 (9)	0.0384 (4)
0.2512 (2)	0.79996 (14)	-0.09487 (12)	0.0495 (5)
0.3777 (3)	0.8243 (2)	-0.1258 (2)	0.0529 (8)
0.3400 (19)	0.8467 (13)	-0.1498 (10)	0.0529 (8)
0.3862 (3)	0.9243 (2)	-0.1444 (2)	0.0508 (8)
0.4378 (17)	0.9049 (12)	-0.1142 (9)	0.0508 (8)
0.3697 (2)	0.98631 (15)	-0.07948 (11)	0.0454 (5)
0.2643 (2)	0.95364 (12)	-0.03269 (9)	0.0359 (4)
0.2056 (2)	1.00408 (12)	0.02394 (8)	0.0333 (4)
0.2411 (2)	1.09378 (12)	0.05058 (9)	0.0366 (4)
0.17003 (14)	1.12454 (10)	0.10892 (7)	0.0360 (3)
0.0741 (2)	1.07456 (12)	0.13839 (9)	0.0345 (4)
0.00326 (14)	1.09584 (11)	0.19227 (8)	0.0425 (4)
-0.07673 (15)	1.02130 (11)	0.20233 (8)	0.0466 (4)
-0.0502 (2)	0.96061 (13)	0.15462 (10)	0.0411 (4)
0.04470 (13)	0.99058 (9)	0.11158 (7)	0.0339 (3)
0.1117 (2)	0.95512 (11)	0.05486 (9)	0.0333 (4)
0.2016 (2)	1.21037 (12)	0.14291 (9)	0.0358 (4)
0.2991 (2)	1.21260 (14)	0.19048 (10)	0.0442 (5)
0.3274 (2)	1.2942 (2)	0.22400 (11)	0.0553 (6)
0.2571 (2)	1.3707 (2)	0.21073 (12)	0.0580 (6)
0.1597 (2)	1.36710 (14)	0.15364 (14)	0.0600 (6)
0.1308 (2)	1.28684 (14)	0.12834 (12)	0.0490 (5)
0.32536 (14)	1.14048 (9)	0.02664 (7)	0.0527 (4)

† Partially occupied (see below).

# Table 2. Selected geometric parameters (Å, °)

S-C15	1.708 (2)	C8N9	1.409 (2)
S-C1	1.743 (2)	N9C10	1.376 (2)
C1C6	1.357 (3)	N9C16	1.453 (2)
C1C2	1.504 (3)	C10-N11	1.302 (2)
C2C3	1.513 (4)	C10-N14	1.369 (2)
C2C3'	1.56 (2)	N11N12	1.401 (2)
C3—C4	1.512 (5)	N12-C13	1.296 (2)
C3'C4'	1.51 (3)	C13—N14	1.370 (2)
C4—C5	1.534 (3)	N14-C15	1.386 (2)
C4′—C5	1.54 (2)	C16C17	1.373 (3)
C5C6	1.506 (3)	C16C21	1.381 (3)

C6C7	1.441 (2)	C17—C18	1.388 (3)
C7-C15	1.364 (2)	C18-C19	1.373 (4)
C7—C8	1.460 (2)	C19—C20	1.367 (4)
C8	1.217 (2)	C20-C21	1.388 (3)
C15SC1	90.10 (8)	C10-N9-C16	117.23 (14)
C6C1C2	125.7 (2)	C8-N9-C16	119.77 (14)
C6C1S	112.76 (13)	N11-C10-N14	111.7 (2)
C2-C1-S	121.54 (15)	N11-C10-N9	128.1 (2)
C1-C2-C3	110.1 (2)	N14-C10-N9	120.14 (15)
C1-C2-C3'	108.3 (7)	C10-N11-N12	105.71 (14)
C4—C3—C2	111.8 (3)	C13-N12-N11	108.12 (15)
C4'—C3'—C2	112.1 (14)	N12-C13-N14	110.5 (2)
C3-C4-C5	112.7 (2)	C10-N14-C13	103.93 (14)
C3'-C4'-C5	107.6 (14)	C10-N14-C15	120.30 (14)
C6C5C4	111.2 (2)	C13-N14-C15	135.7 (2)
C6C5C4'	110.6 (6)	C7-C15-N14	120.6 (2)
C1-C6-C7	111.9 (2)	C7-C15-S	113.94 (13)
C1-C6-C5	121.8 (2)	N14-C15-S	125.40 (13)
C7—C6—C5	126.4 (2)	C17-C16-C21	121.6 (2)
C15-C7-C6	111.3 (2)	C17-C16-N9	118.9 (2)
C15-C7-C8	121.3 (2)	C21-C16-N9	119.4 (2)
C6-C7-C8	127.3 (2)	C16C17C18	118.8 (2)
O22-C8-N9	120.4 (2)	C19-C18-C17	120.3 (2)
O22-C8-C7	125.0 (2)	C20-C19-C18	120.1 (2)
N9-C8-C7	114.67 (15)	C19-C20-C21	120.8 (2)
C10-N9-C8	122.92 (15)	C16C21C20	118.3 (2)
C1-C2-C3-C4	-44.7 (3)	C1-C2-C3'-C4'	45.6 (17)
C2-C3-C4-C5	61.2 (4)	C2—C3′—C4′—C5	-67.5 (19)
C3-C4-C5-C6	-43.0 (4)	C3'-C4'-C5-C6	52.5 (16)
C4-C5-C6-C1	13.0 (3)	C4'-C5-C6-C1	-21.0 (8)
C5-C6-C1-C2	0.6 (3)	C6C1C2C3'	-12.0 (9)
C6-C1-C2-C3	15.3 (3)		

Disorder of atoms C3 and C4 was observed after anisotropic refinement of the non-H atoms. C3 and C4 were assigned occupancies of p and C3' and C4' were assigned occupancies of 1-p, with p = 0.850 (6) at convergence. All the H atoms belonging to the major conformer were located from difference Fourier maps and refined isotropically.

Data collection: XSCANS (Siemens, 1994). Cell refinement: XSCANS. Data reduction: XSCANS. Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990a). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: SHELXTL/PC (Sheldrick, 1990b). Software used to prepare material for publication: SHELXL93. Geometric calculations: PARST (Nardelli, 1983).

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Lists of structure factors, anisotropic displacement parameters, Hatom coordinates and complete geometry have been deposited with the IUCr (Reference: MU1178). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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# Disordered Structure of 2-Methylanthraquinone

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## Abstract

Crystals of the title compound,  $C_{15}H_{10}O_2$ , are monoclinic, space group  $P2_1/c$ . The molecules are disordered around a centre of symmetry. Packing considerations and the calculation of the intermolecular interaction energies show that the disordered structure has occurred by random packing of structures with space groups  $P2_1$ and Pc, the maximal subgroups of  $P2_1/c$ .

#### Comment

A certain kind of structural disorder has been observed in homomolecular crystals like those of azulene (Robertson, Shearer, Sim & Watson, 1962) and 1-bromo-4chlorobenzene (Klug, 1947). In these crystals, each molecule is apparently disordered in two orientations related by a  $\overline{1}$  symmetry operation. Here, we comment on such a structural disorder as found in 2-methylanthraquinone, (I).



The molecular arrangement is characterized by the formation of a molecular sheet in  $(10\overline{2})$  and molecular stacking along the *a* axis with an interplanar spacing